

#### Marco Island, Florida

Water Quality

Hypoxia Investigation

8.25.21

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## Summary

- **Dissolved Oxygen (DO) deficiency** has been observed within the Rookery Bay estuary, centered on Marco Island - threat to aquatic life
- **DO "pit" and pH "dome"** have formed over Marco (only) since 2020
- Marco "flirting" with hypoxia DO and pH trends variable & negative
- *Hypothesis*: Nutrients feed Harmful Algal Blooms (HAB) which deplete oxygen leading to seagrass die-off and manatee mortality
- *Reuse water TN and TP* appear to be nutrient "food" sources for HAB which depletes DO and spikes pH
- Marco impaired for TN If current trends persist, anticipate DO impairment in 2023
- Basins 1,2 and 4 are impaired for DO; Swallow location (Cape Marco Point) is now hypoxic
- Basin 5 is healthy; Basin 5 gets NO reuse water

Marco Island waterways are impaired for dissolved oxygen

## Background

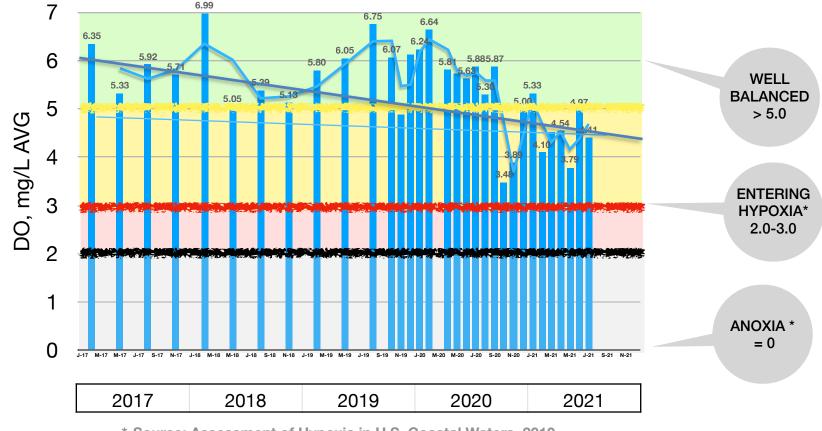
- Water quality on Marco has been a concern for many years Highlights:
  - 10/9/13: Nancie Ritchie "Our Canals Need Care & Conservation"
  - 9/14/14: Phares Hindl Memo on need for Waterway Management Plan
  - 01/4/17: Dr. W.D. Trotter White paper recommends monthly water testing
  - Circa 2018: Position Paper by Sam Young
  - 4/29/19: City Council Convenes 2<sup>nd</sup> Water Quality Workshop
  - 8/22/19: FDEP issues TN impairment letter

• Much has been done:

- 2006-2012 STRP Septic Tank Replacement Program
- 2016 Fertilizer Ordinance
- 2019 50% Reduction of TN in reuse water by WWTP
- These efforts have not resolved the issue

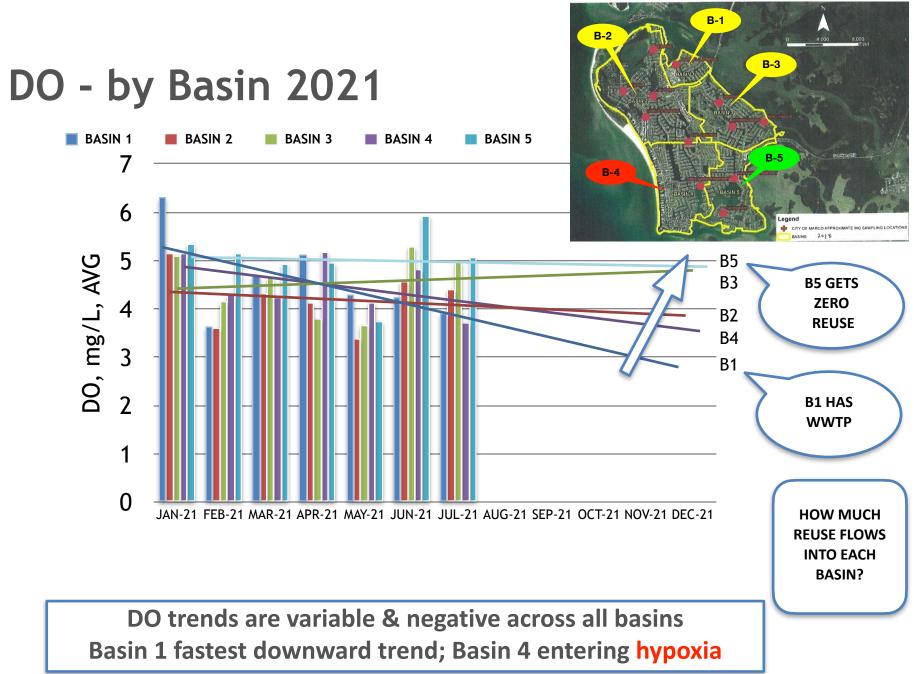
Water quality on Marco Island continues to deteriorate

#### DO Trend - MARCO 2017-2021

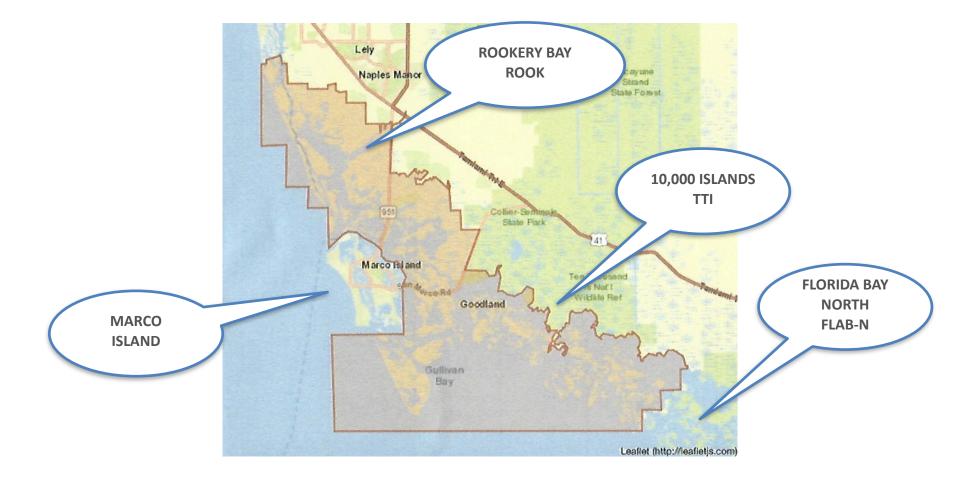


\* Source: Assessment of Hypoxia in U.S. Coastal Waters, 2010 \* Source: City of Marco Island Water Quality Sampling Reports

#### Marco is "flirting" with hypoxia If trend continues - anticipate Marco DO impairment in 2023

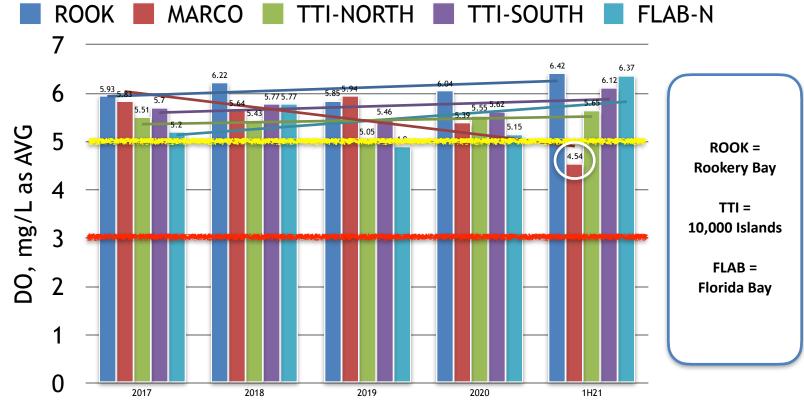


#### **Rookery Bay Estuary**



Marco is in the middle of an estuarine eco-system

#### Estuary Dissolved Oxygen (DO)



Source: Marco (2018-2021) City of Marco Island Water Quality Sampling Reports Source: Marco (2017) WIN Source: Rook & TTI - DBHYDRO

Estuary DO improving - Marco DO declining!

#### **Estuary Dissolved Oxygen - 1H21**

Oxygen, Dissolved (DO) Milligrams per liter (mg/L)

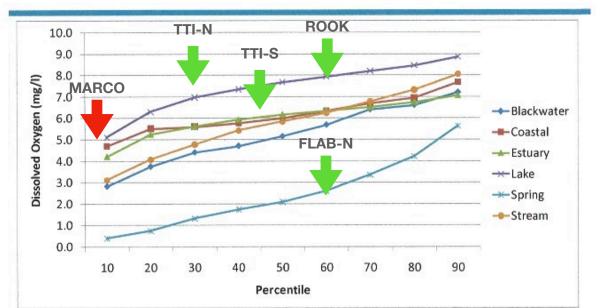
Dissolved oxygen (DO), which is the amount of free (not chemically combined) oxygen dissolved in water or wastewater, is an essential component in the aquatic environment. The most important and commonly used measurement of water quality, it indicates a waterbody's state of health—that is, the ability to support desirable aquatic life. A vast array of aquatic organisms depends on the presence of adequate levels of DO for survival. DO concentrations in a waterbody vary seasonally.<sup>3</sup>

Generally, waters with DO concentrations of 5.0 mg/L or higher can support a wellbalanced, healthy biological community. However, some species cannot tolerate even slight depletion, and when DO concentrations fall below natural levels, the result is often a complete alteration of the community structure. The consequences of these changes often have both ecological and economic significance.

Some systems with "good" water quality exhibit naturally low DO concentrations (e.g., swamps). Also, daytime and mean measurements of DO have limited significance, since nocturnal respiration and other episodic instances of low oxygen can significantly affect aquatic life. Diurnal DO studies are important to understanding the oxygen cycle in a particular waterbody.

Source: NALMS





Percentile	Blackwater	Coastal	Estuary	Lake	Spring	Stream
10	2.8	4.7	4.2	5.1	0.4	3.1
20	3.7	5.5	5.2	6.3	0.7	4.1
30	4.4	5.6	. 5.6	7.0	1.3	4.8
40	4.7	5.8	5.9	7.4	1.7	5.4
50	5.1	6.0	6.2	7.7	2.1	5.8
60	5.7	6.3	6.4	7.9	2.6	6.2
70	6.4	6.7	6.5	8.2	3.3	6.8
80	6.6	6.9	6.7	8.4	4.2	7.3
90	7.2	7.7	7.1	8.9	5.6	8.0
# of waterbodies	56	92	448	833	100	1394



🥚 < 3 mg/L

< 5 mg/L</p>

MO-YR	BARF	OLDE	B-1	JHPA	KEND	COLL	HCCE	B-2	HOLL	нимм	WIND	B-3	LAND	SWAL	WWIN	B-4	EWIN	MCIL	B-5	MARCO
Jan-17																				
Feb-17	6.82		6.82	6.51	5.82	6.12	6.89	6.34	6.08	6.25	6.10	6.14			6.47	6.47	6.44		6.44	6.35
Mar-17																				
Apr-17																				
May-17	5.44		5.44	5.48	5.36	5.07	4.55	5.12	5.26	5.53	4.75	5.18			6.03	6.03	5.79		5.79	5.33
Jun-17																				
Jul-17																				
Aug-17	5.75		5.75	5.83	5.87	6.22	6.16	6.02	5.67	6.24	5.40	5.77			6.11	6.11	5.97		5.97	5.92
Sep-17																				
Oct-17																				
Nov-17	6.07		6.07	5.78	5.52	5.87	5.29	5.62	5.61		6.16	5.89			5.47	5.47	5.62		5.62	5.71
Dec-17																				
AVG	6.02		6.02	5.90	5.64	5.82	5.72	5.77	5.66	6.01	5.60	5.74			6.02	6.02	5.96		5.96	5.83

2 "impaired" readings; 4% of all observations

🔵 < 3 mg/L

< 5 mg/L</p>

MO-YR	BARF	OLDE	B-1	JHPA	KEND	COLL	HCCE	B-2	HOLL	нимм	WIND	B-3	LAND	SWAL	WWIN	B-4	EWIN	MCIL	B-5	MARCO
Jan-18																				
Feb-18	6.88		6.88	6.94	7.52	7.37	7.44	7.32	5.87		6.69	6.28			7.05	7.05	7.30	6.83	7.07	6.99
Mar-18																				
Apr-18																				
May-18	5.16		5.16	4.86	4.56	4.87	5.51	4.95	3.97	4.54	4.72	4.41			6.73	6.73	5.20	5.45	5.33	5.05
Jun-18																				
Jul-18																				
Aug-18	6.67		6.67	5.13	4.98	4.64	5.03	4.95	4.69	4.89	5.81	5.13			5.92	5.92	5.80	5.74	5.77	5.39
Sep-18																				
Oct-18																				
Nov-18	5.65		5.65	4.71	5.00	5.73	5.86	5.33	4.20	4.88	5.29	4.79			5.30	5.30	4.73	5.09	4.91	5.13
Dec-18																				
AVG	6.09		6.09	5.41	5.52	5.65	5.96	5.63	4.68	4.77	5.63	5.15			6.25	6.25	5.76		5.77	5.64

13 "impaired" readings; 24% of all observations

) < 3 mg/L 🔵 < 5 mg/L

MO-YR	BARF	OLDE	B-1	JHPA	KEND	COLL	HCCE	B-2	HOLL	нимм	WIND	B-3	LAND	SWAL	WWIN	B-4	EWIN	MCIL	B-5	MARCO
Jan-19																				
Feb-19	6.23		6.23	5.50	5.15	5.77	6.58	5.75	5.08	6.22	5.97	5.76			5.78	5.78	5.50	5.98	5.74	5.80
Mar-19																				
Apr-19																				
May-19	6.09		6.09	5.70	5.48	7.04	6.18	6.10	5.45	6.37	5.60	5.81			6.44	6.44	6.12	6.06	6.09	6.05
Jun-19																				
Jul-19																				
Aug-19	6.09		6.09	7.32	6.72	9.23	10.36	8.41	4.97	5.55	4.74	5.09			6.83	6.83	6.13	6.28	6.21	6.75
Sep-19																				
Oct-19	6.64	4.73	5.69	6.05	5.14	6.18	7.94	6.33	4.34	6.79	5.57	5.57	5.03	7.52	5.69	6.08	6.68	6.70	6.69	6.07
Nov-19	6.05	5.12	5.59	4.58	4.19	4.25	5.67	4.67	4.90	4.30	5.94	5.05	4.12	3.80	4.74	4.22	5.43	5.29	5.36	4.88
Dec-19	6.53	6.92	6.73	5.53	4.88	5.92	6.73	5.77	5.16	5.35	6.96	5.82	6.89	6.33	6.63	6.62	5.85	5.93	5.89	6.12
AVG	6.27	5.59	6.07	5.78	5.26	6.40	7.24	6.17	4.98	5.76	5.80	5.51	5.35	5.88	6.02	5.99	5.95	6.04	6.00	5.94

13 "impaired" readings; 17% of all observations

😑 < 3 mg/L

○ < 5 mg/L</p>

MO-YR	BARF	OLDE	B-1	JHPA	KEND	COLL	HCCE	B-2	HOLL	нимм	WIND	B-3	LAND	SWAL	WWIN	B-4	EWIN	MCIL	B-5	MARCO
Jan-20	6.30	6.13	6.22	6.26	6.26	6.71	6.26	6.37	5.36	6.33	6.27	5.99	6.93	5.52	6.22	6.22	6.79	6.04	6.42	6.24
Feb-20	6.57	7.30	6.94	6.52	6.63	7.59	6.64	6.85	5.32	6.77	6.52	6.20	7.72	5.42	6.38	6.51	6.37	7.20	6.79	6.64
Mar-20																				
Apr-20	6.43	5.50	5.97	5.23	5.39	4.52	4.81	4.99	5.73	5.37	5.34	5.48	6.42	6.58	5.55	6.18	6.05	5.55	5.80	5.61
May-20	6.17	6.13	6.15	6.13	5.55	5.27	5.50	5.61	5.42	5.94	5.79	5.72	5.22	5.52	5.78	5.51	5.76	6.25	6.01	5.75
Jun-20	5.81	5.08	5.45	5.27	4.88	5.26	5.29	5.18	6.03	6.87	7.47	6.79	6.66	1.65	6.39	4.90	6.23	5.86	6.05	5.63
Jul-20	5.24	4.79	5.02	5.51	5.13	5.65	6.41	5.68	4.90	6.17	5.64	5.57	7.59	6.20	7.06	6.95	6.16	5.92	6.04	5.88
Aug-20	5.92	5.20	5.56	5.09	4.99	4.26	4.33	4.67	4.99	4.71	5.32	5.01	6.52	6.31	5.33	6.05	5.48	5.75	5.62	5.30
Sep-20	6.00	5.69	5.85	5.66	6.01	5.75	5.57	5.75	5.75	6.29	6.50	6.18	5.83	5.59	5.90	5.77	6.17	5.40	5.79	5.87
Oct-20	3.80	3.63	3.72	3.99	3.66	3.54	2.90	3.52	3.80	3.00	3.50	3.43	2.46	3.75	4.63	3.61	2.82	3.18	3.00	3.48
Nov-20	4.80	3.75	4.28	4.44	3.60	3.67	4.48	4.05	3.47	3.60	3.90	3.66	3.68	2.76	4.13	3.52	4.20	3.97	4.09	3.89
Dec-20	5.72	5.43	5.58	6.21	4.97	4.82	5.16	5.29	5.17	4.91	4.72	4.93	3.30	3.69	5.20	4.06	5.45	5.21	5.33	5.00
AVG	5.71	5.33	5.52	5.48	5.19	5.19	5.21	5.27	5.09	5.45	5.54	5.36	5.67	4.82	5.69	5.39	5.59	5.48	5.54	5.39

44 "impaired" readings; 28% of all observations

Marco % Impaired Readings - DO

#### Marco DO Measurements 2021



😑 < 3 mg/L

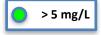
😑 < 5 mg/L

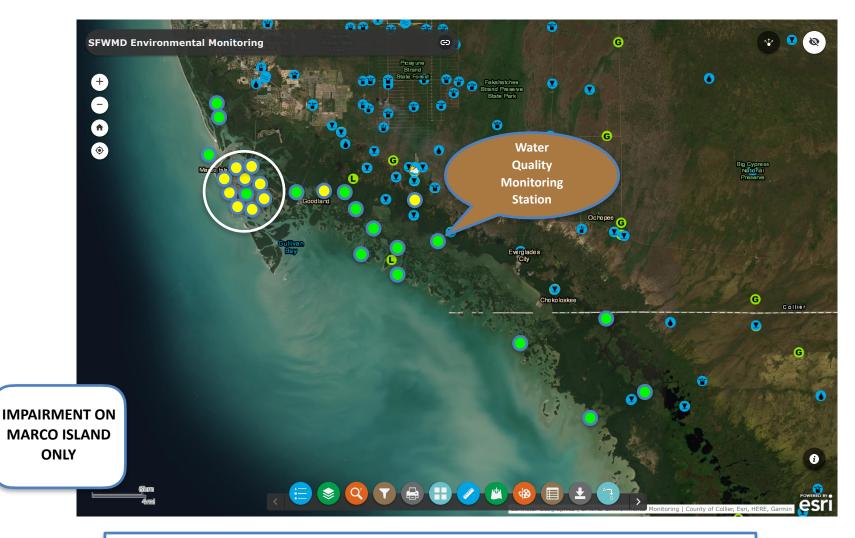
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Jan-21	6.74	5.90	6.32	6.26	4.44	4.91	4.95	5.14	4.77	5.35	5.17	5.10	4.72	4.84	5.92	5.16	5.71	4.99	5.35	5.33
Feb-21	3.60	3.64	3.62	3.68	3.88	3.29	3.57	3.61	4.36	3.67	4.43	4.15	5.90	2.75	4.24	4.30	5.47	4.85	5.16	4.10
Mar-21	5.45	3.91	4.68	5.16	4.73	3.27	4.06	4.31	4.73	4.44	4.80	4.66	5.05	3.58	4.06	4.23	4.90	4.96	4.93	4.51
Apr-21	5.00	5.21	5.11	4.31	4.42	3.94	3.85	4.13	3.70	3.39	4.31	3.80	5.69	4.29	5.58	5.19	4.81	5.09	4.95	4.54
May-21	4.52	4.08	4.30	3.85	3.49	3.17	3.01	3.38	3.68	3.27	4.05	3.67	3.72	3.69	5.01	4.14	3.70	3.78	3.74	3.79
Jun-21		4.25	4.25	5.12	3.87	4.19	5.07	4.56	5.25	5.99	4.63	5.29	6.13	2.66	5.70	4.83	6.74	5.07	5.91	4.97
Jul-21	4.10	3.74	3.92	4.97	3.95	4.47	4.21	4.40	4.82	5.28	4.87	4.99	4.47	1.84	4.81	3.71	5.15	5.01	5.08	4.41
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Dec-21																				
AVG	4.90	4.39	4.60	4.76	4.11	3.89	4.10	4.22	4.47	4.48	4.61	4.52	5.10	3.38	5.05	4.51	5.21	4.82	5.02	4.52

68 "impaired" readings; 69% of all observations Swallow sampling location (Cape Marco Point) Hypoxic

#### Estuary DO Status MAY-21

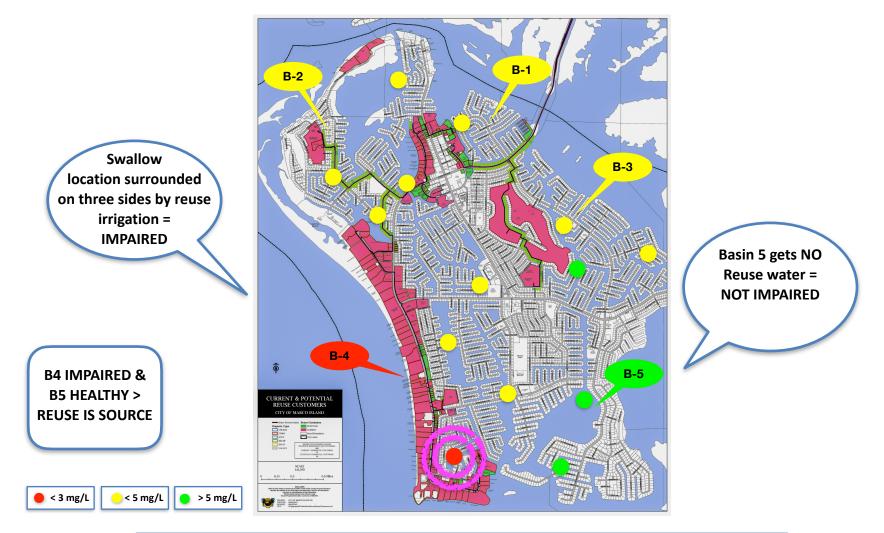






DO "pit" has formed only over Marco Island

#### MARCO DO Status JUL-21



11 locations Impaired; Swallow location Hypoxic

### Monitoring Estuaries, NOAA

- "Oxygen depletion may occur in estuaries when many plants die and decompose, or when wastewater with large amounts of <u>organic</u> material enters the estuary"
  - 99% of the TN in Marco waterways is <u>organic</u> (99% TKN JUN-21)
- "In some estuaries, large nutrient inputs, typically from sewage, stimulate algal blooms"
  - 100% of the reuse water TN is <u>organic</u> (84% NO3; 16% TKN JUN-21)
- "When the algae die, they begin to decompose"
- "The process of decomposition depletes the surrounding water of oxygen and, in severe cases, leads to hypoxic (very low oxygen) conditions that kill aquatic animals"

Marco has large amounts of <u>organic</u> material in the waterways

## Root Cause - Nutrients Feed Algal Blooms

- "Much of the resulting nitrogen and phosphorus that feeds algal blooms comes from fertilizer runoff and sewage leaks" Naples News, 7/12/21
- "Blooms kill seagrass manatee's main food source by robbing the water of oxygen and sunlight the grass needs to grow" Naples News 7/12/21
- Sources of Nutrients on Marco Island:
  - TN: Reuse water, fertilizer, stormwater
  - TP: Reuse water the only source on the island?

Nutrients feed algal blooms, which consume oxygen and block sunlight, killing seagrass and leading to manatee mortality

#### "Nitrogen and Phosphorus in Reuse and Potable Water", 2018 - position paper by Sam Young

<ul> <li>Annual nitrogen load from reuse and potable water</li> </ul>	88,564 #
• Annual phosphorus load from reuse and potable water (2)	36,783 #
<ul> <li>Nitrogen load equivalent annual 20# bags of fertilizer</li> </ul>	42,782
• Daily 20# bags of nitrogen fertilizer equivalent into Marco waterbody (3)	171

(1) Logistics: Hire 2-3 full-time employees; buy 2 barges; build a dock, build fertilizer warehouse; maintain 30 day inventory (4,000 bags)

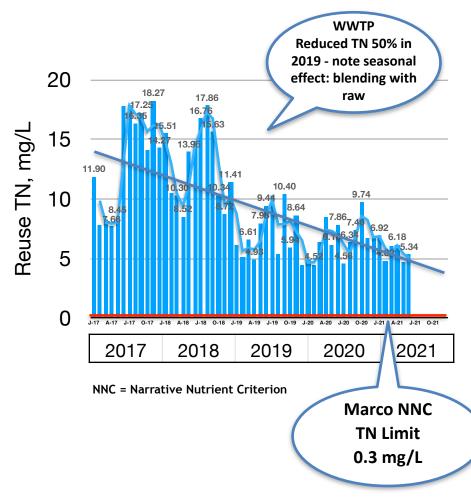
(2) We do not allow ANY TP in our fertilizer - yet we spread 1.5 tons PURE TP each month with reuse water

(3) 5 days per week; 50 weeks per year

WWTP has since dropped TN by 50%; TP remains about the same

Imagine if city staff had to spread this much fertilizer - a lot of work! (1)

### Marco Island WWTP - Reuse TN



"<u>The major sources of excessive amounts</u> of nitrogen in surface water are the effluent from municipal treatment plants and runoff

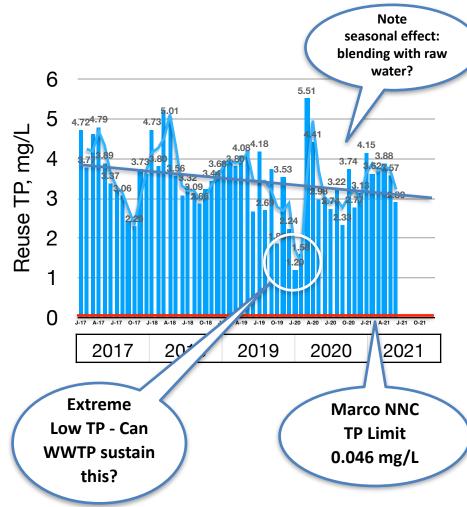
from agricultural sites. When nutrient concentrations consistently exceed natural levels, the resulting nutrient imbalance can cause undesirable changes in a waterbody's biological community and <u>increase the rate of</u> <u>eutrophication</u> (or aging) in an aquatic system. Usually, the eutrophication process is observed

as a change in the structure of the algal community and includes severe algal blooms that may cover large areas of a waterbody for extended periods. <u>Large blooms are generally</u> <u>followed by a depletion in dissolved oxygen</u> <u>concentrations as a result of algal decomposition</u>"

**FDEP** 

At the 2017 peak, TN level was 61X the NNC - now only 18X

### Marco Island WWTP - Reuse TP

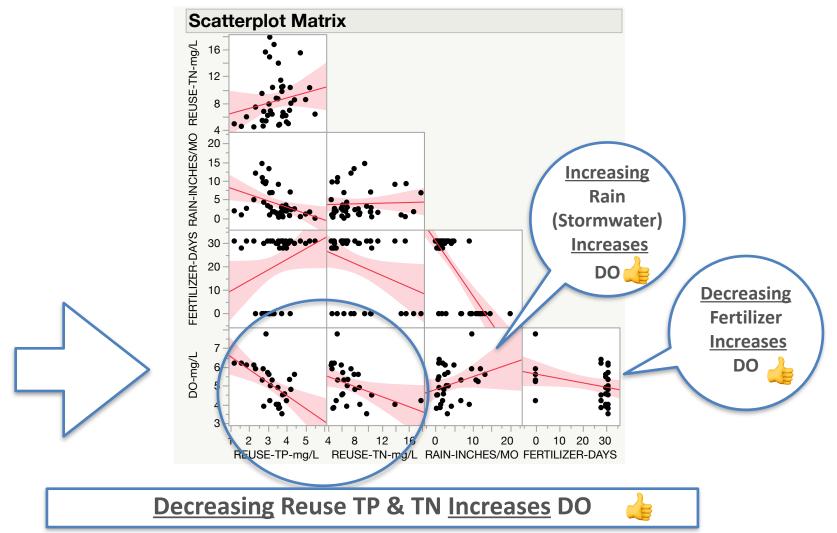


"<u>High phosphorus concentrations are frequently</u> <u>responsible for accelerating the process of</u> <u>eutrophication</u> (or aging) of a waterbody. Once phosphorus and other important nutrients enter the ecosystem, they are extremely difficult to remove because they are taken up by plants or <u>deposited in sediments</u>. Nutrients, particularly phosphates, deposited in sediments generally are redistributed into the water. <u>This type of</u> <u>cycling compounds the difficulty of halting the</u> <u>eutrophication process</u>"

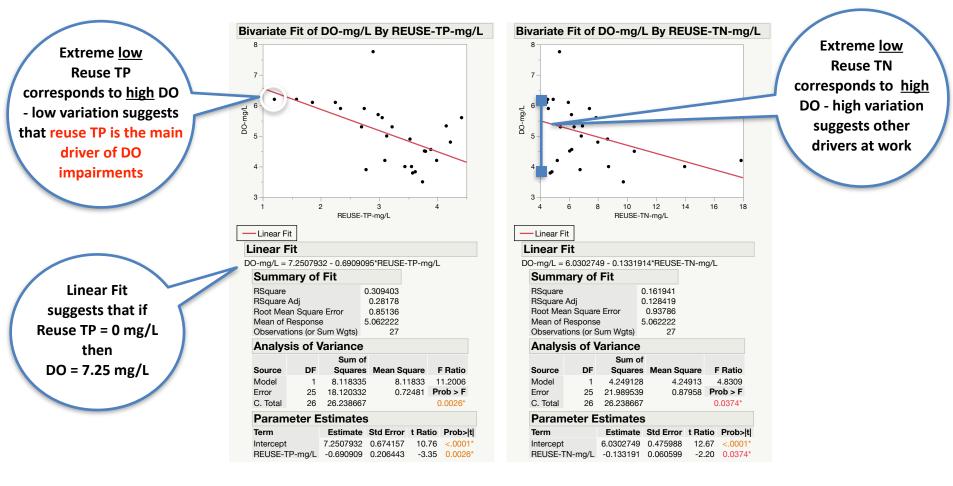
FDEP

At the 2020 peak, TP level was 120X the NNC - now only 63X

### Marco Water Quality Database: 5 YR by MO

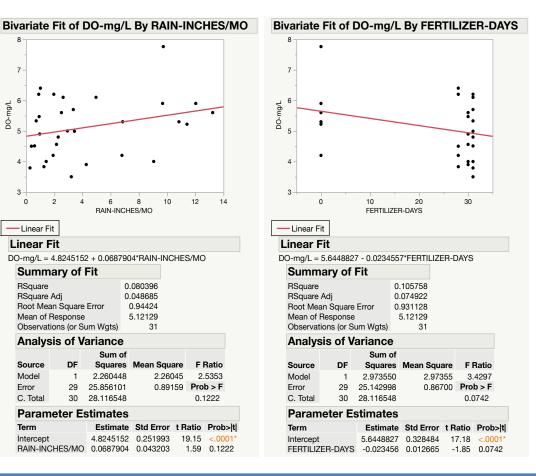


#### DO By Reuse-TP & Reuse-TN



Decreasing Reuse TP & TN increases DO reducing TP is the next *MAJOR* hurdle for Marco WWTP

### DO By Rain & Fertilizer Days

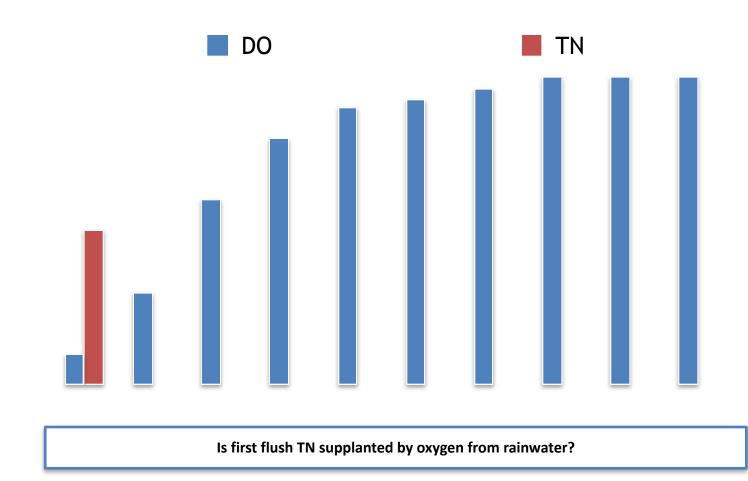


**Increasing Rain increases DO** 



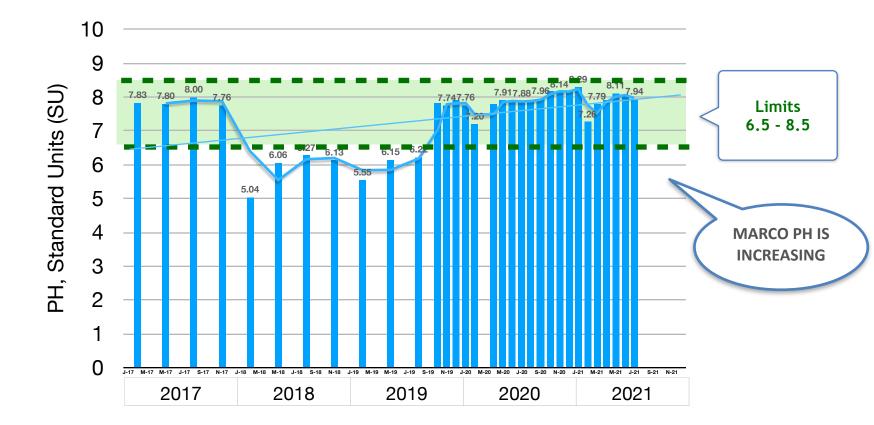
**Decreasing Fertilizer increases DO** 

### First Flush Hypothesis - Stormwater



Marco Waterways NEED this source of oxygen

#### pH Trend - More Evidence



"An overabundance of algae (called an algal bloom) may cause pH levels in an estuary to rise significantly, and this can be lethal to aquatic animals" NOAA

#### pH (PH) Standard units (SU)

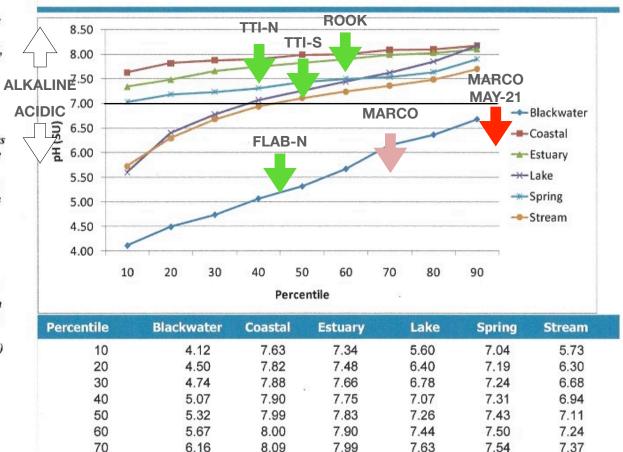
The pH of a body of water denotes its hydrogen ion activity, based on the negative logarithm of hydrogen ion concentrations. A pH of 1 to 7 is acidic, a pH of 7 is neutral, and a pH of 7 to 14 is alkaline.

pH significantly affects chemical and biological interactions in the aquatic environment. This is of particular concern in considering the effects of toxic substances on aquatic organisms, especially the release of metals from sediments. At certain pH levels, a particular toxicant may increase in toxicity or become more soluble, and thus is more likely to affect aquatic organisms.

The problems of acidic deposition and the acidification of lakes and streams have gained widespread attention. However, certain biological communities are adapted to acidic conditions (e.g., blackwater stream systems, where pH ranges from 4 to 5) or to slightly alkaline conditions (e.g., spring runs, where pH values of 8 are not unusual) and are endangered only when the natural conditions are altered.

Source: FDEP

#### Estuary pH - 1H21



8.03

8.10

447

8.10

8.18

81

7.86

8.18

836

7.64

7.90

100

1H21 Marco pH was higher than 70% of Florida estuaries May-21 Marco pH was higher than 95% of Florida estuaries

6.37

6.68

56

80

90

# of waterbodies

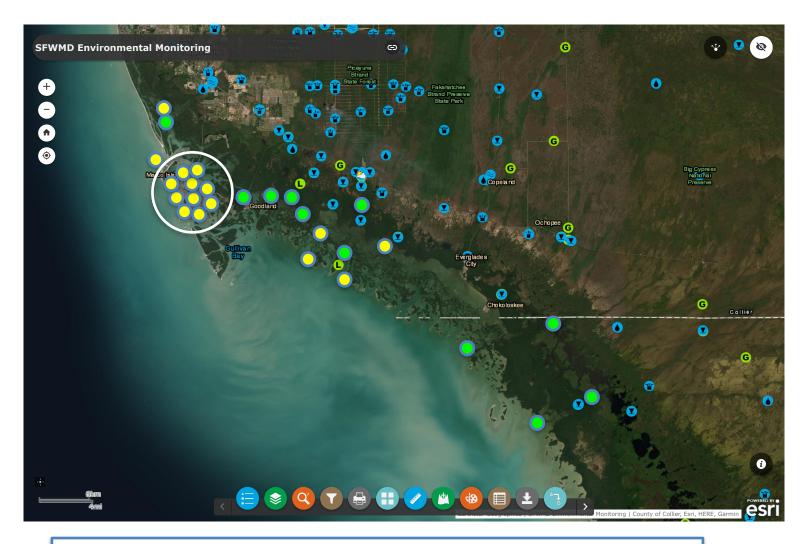
7.49

7.70

1397

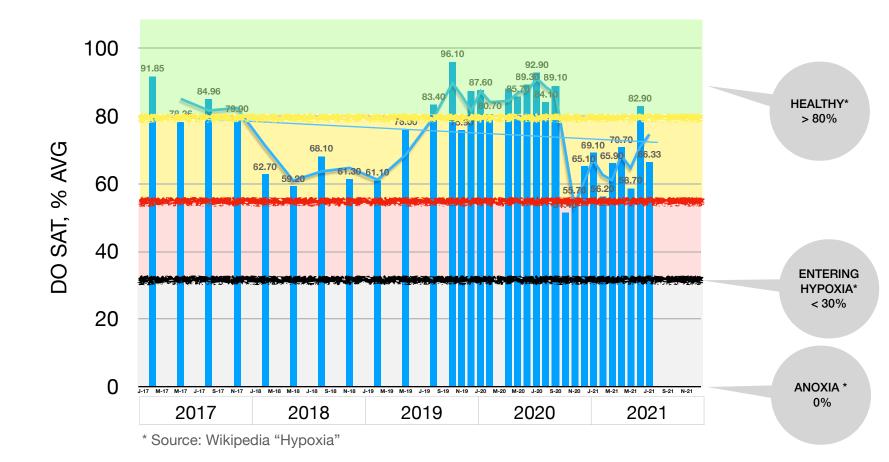
### Estuary pH Status MAY-21





pH "dome" has formed only over Marco Island

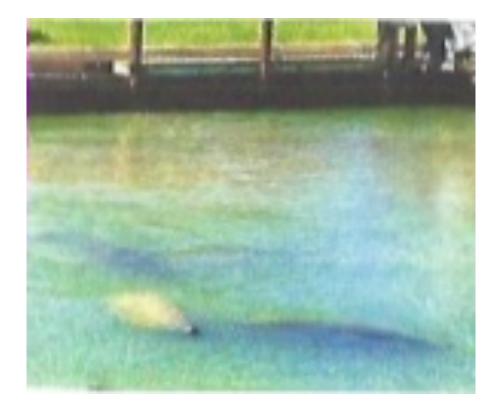
#### DOSAT % Trend - even more evidence



Marco DOSAT trending to Hypoxia - other parameters may indicate Hypoxia as well

#### Impact on Manatee - Naples Daily News 7/12/21

- "More manatees have died in 2021 than any other year"
- "841 manatees have died in Florida waters, mostly in Brevard County"
- No real manatee death on Marco? Maybe they just do not come here anymore? They can find other food sources in the estuary
- Brevard County the manatee may have nowhere to go and die of starvation



Can we track the manatee population like we do dolphin? When was the last time the water looked like this photo on Marco?

## Hypoxia (environmental)

#### Causes of Hypoxia

"Oxygen depletion can result from a number of natural factors, but is most often a concern as a consequence of pollution and eutrophication in which plant nutrients enter a river, lake, or ocean, and phytoplankton blooms are encouraged"

"As phytoplankton break down, free phosphorus and nitrogen become available in the environment, which also fosters hypoxic conditions"

#### Solutions

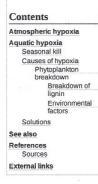
"To combat hypoxia, it is essential to reduce the amount of land-derived nutrients reaching rivers in runoff. This can be done by improving sewage treatment and by reducing the amount of fertilizers leaching into the rivers"

"Other natural habitat-based solutions include restoration of shellfish populations, such as oysters"

#### WIKIPEDIA

#### Hypoxia (environmental)

Hypoxia refers to low oxygen conditions. Normally, 20.9% of the gas in the atmosphere is oxygen. The partial pressure of oxygen in the atmosphere is 20.9% of the total barometric pressure.<sup>[3]</sup> In water, oxygen levels are much lower, approximately 7 ppm 0.0007% in good quality water, and fluctuate locally depending on the presence of photosynthetic organisms and relative distance to the surface (if there is more oxygen in the air, it will diffuse acrossure gradient).<sup>[4]</sup>





Global areas of ocean hypoxia 2009 Global map of low and declining oxygen levels in the open ocean and coastal waters.<sup>10</sup> The map indicates coastal sites where anthropogenic nutrients have exacerbatied or caused oxygen declines to <2 mgl (<63 µmol/l) (red dots), as well as ocean <u>oxygen</u> minimum zones at 300 m (blue shader regions).<sup>20</sup>

#### Atmospheric hypoxia

Atmospheric hypoxia occurs naturally at high altitudes. Total atmospheric pressure decreases as altitude increases, causing a lower partial pressure of oxygen which is defined as hypobaric hypoxia. Oxygen remains at 20.9% of the total gas mixture, differing from hypoxic hypoxia, where the percentage of oxygen in the air (or blood) is decreased. This is common in the sealed burrows of some subterranean animals, such as blesmols.<sup>[5]</sup> Atmospheric hypoxia is also the basis of altitude training which is a standard part of training for elite athletes. Several companies minic hypoxia using normobaric artificial atmosphere.

Hypoxia appears to be a self-sustaining, accelerating process

### **Managed Nutrient Reduction**

"The Rhode Island Department of Environmental Management (RI DEM) implemented nitrogen reductions in waste water treatment facilities (WWTF) beginning in 2005"

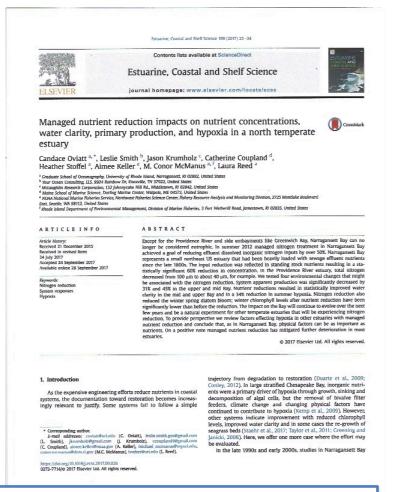
(RI DEM, 2005)

"By summer 2012 the overall goal of a 50% reduction of the WWTP effluent DIN load was achieved in Narragansett Bay"

(Liberti, A. 2014 RI DEM. personal communication)

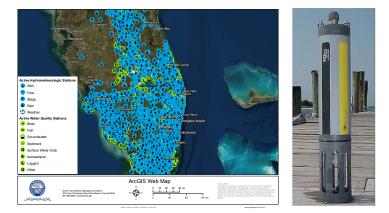
"Management regulations have mandated tertiary treatment at WWTP year round ..."

DIN = Dissolved Inorganic Nitrogen



Could the City commission a study like this for the Marco waterways?

#### **Next Steps**



- 1. Continue research into the Root-Cause of variation in Marco waterbody chemistry: DO, TN, TP, CHLA, pH; produce quarterly reports; examine variation across the estuary for all 17 parameters that are measured
- 2. Investigate cost of Tertiary Treatment of reuse water to lower the reuse TN & TP levels to NNC standards: TN = 0.3 mg/L; TP = 0.046 mg/L <or> to sustainable levels, known as "Managed Nutrient Reduction"
- 3. Evaluate possibility of getting Marco WQ data into DBHYDRO for ease of data analysis current data in WIN CANNOT be accessed
- 4. Evaluate cost of continuous-automatic Data Loggers for Marco Island and nearby waterbodies: Data logged each 30 minutes Depth, Salinity, DO, Turbidity, pH critical eutrophication parameters; "Mirror" the NOAA SWMP System
- Investigate "Developing and Implementing an Estuarine Water Quality Monitoring, Assessment, and Outreach Program" - The MYSound Project (See City Council Water Quality Workshop 4/29/19)

Solution: Clean up the reuse water

## Motions for consideration: WAC 8.19.21

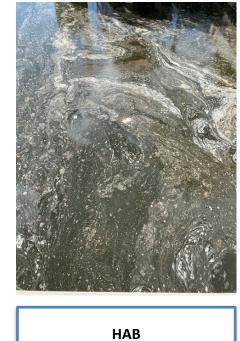
- 1. Post the documents from today's agenda on the WAC web page for public access:
  - A. "Marco Hypoxia Investigation"
  - B. "Managed Nutrient Reduction", Estuarine, Coastal and Shelf Science
  - C. "Hypoxia (environmental)", Wikipedia
- 2. Investigate the cost of tertiary treatment of reuse water to lower the reuse TN & TP levels to NNC standards: TN = 0.3 mg/L; TP = 0.046 mg/L or to sustainable levels, known as "Managed Nutrient Reduction".
- 3. Evaluate the possibility of getting Marco WQ data into DBHYDRO for ease of data analysis.
- 4. Assess the cost of continuous-automatic data loggers for Marco Island and nearby waterbodies: Data logged each 30 minutes Depth, Salinity, DO, Turbidity, pH critical eutrophication parameters; "Mirror" the NOAA SWMP system.
- 5. Investigate "Developing and Implementing an Estuarine Water Quality Monitoring, Assessment, and Outreach Program" The MYSound Project (See City Council Water Quality Workshop 4/29/19).
- 6. Commission a study that investigates restoration of the Marco waterbody shellfish and seagrass populations with recommendations and implementation plans.
- 7. Implement a demonstration project for the installation of (1) compressed air injection system at the Swallow water quality sampling station to determine the effectiveness of this technology in increasing DO in the waterway.
- 8. Commission a study (See "Managed Nutrient Reduction") to understand the effects of any actions taken to resolve the hypoxic conditions in the Marco Island waterways.

#### Motions not voted

### **Questions?**

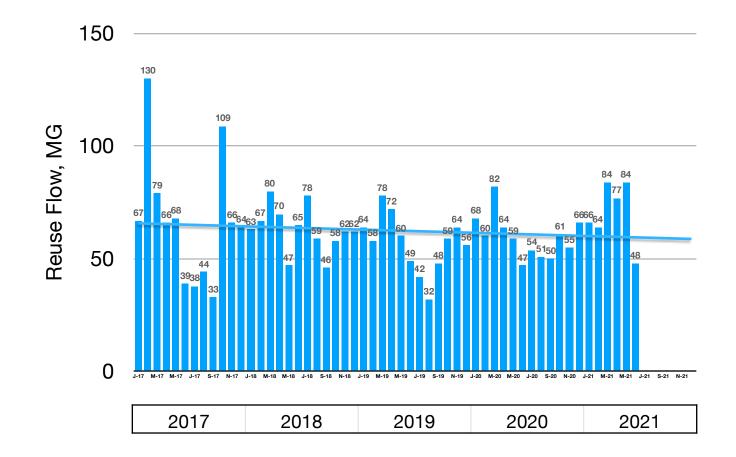






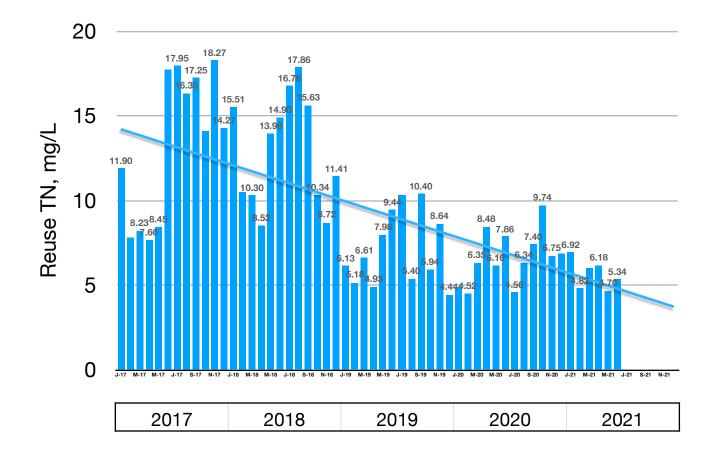
# Appendix

#### Marco Island WWTP - Reuse Flow



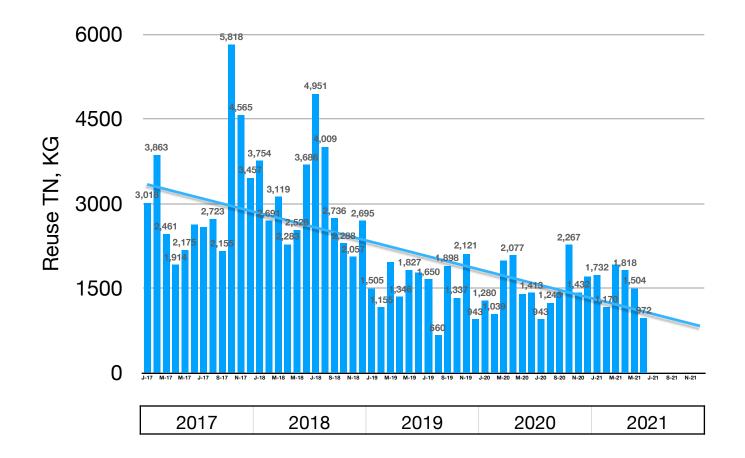
Reuse flow is seasonal - dropping during the off-season

#### Marco Island WWTP - Reuse TN



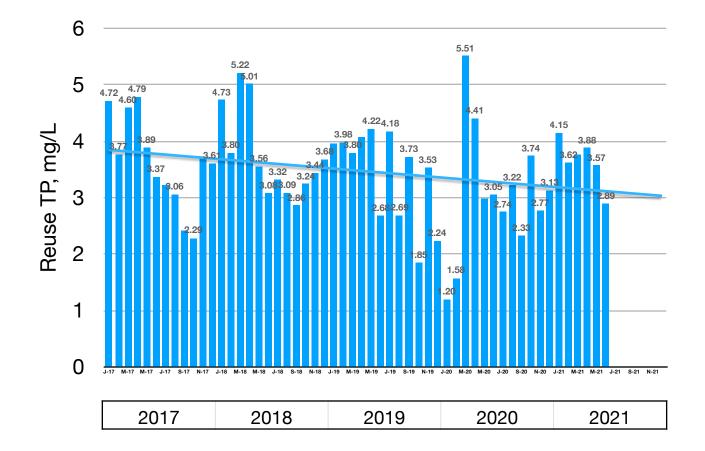
Reuse TN was reduced by 50% in early 2019 - 1H21 Average = 5.66 mg/L

#### Marco Island WWTP - Reuse TN Load



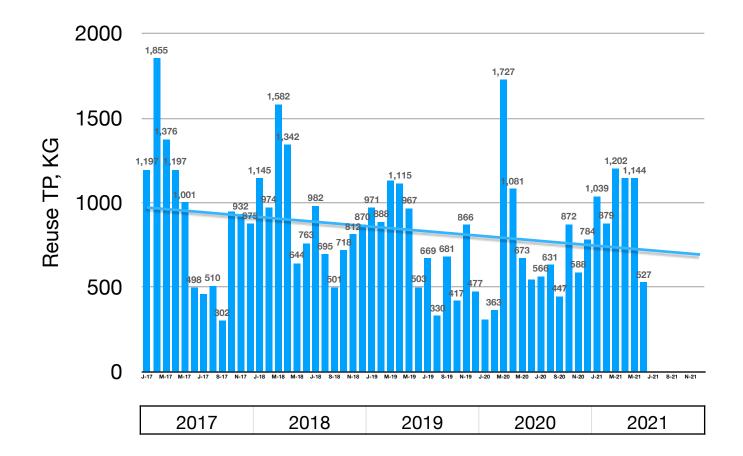
Reuse TN load reduced significantly in early 2019

#### Marco Island WWTP - Reuse TP



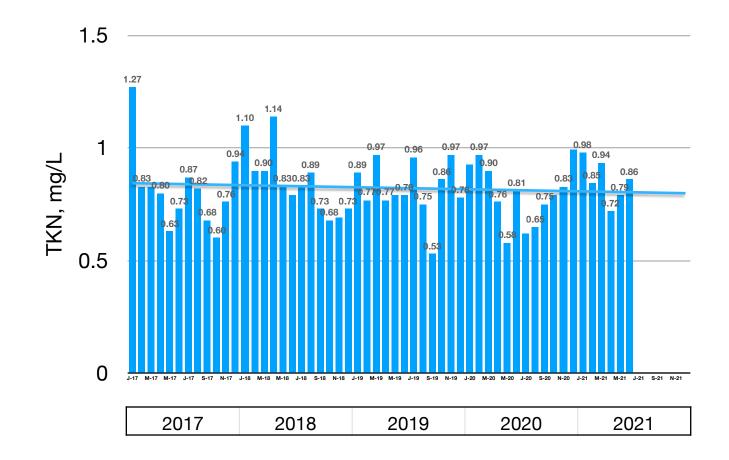
TP highly variable - 1H21 Average = 3.64 mg/L

### Marco Island WWTP - Reuse TP Load



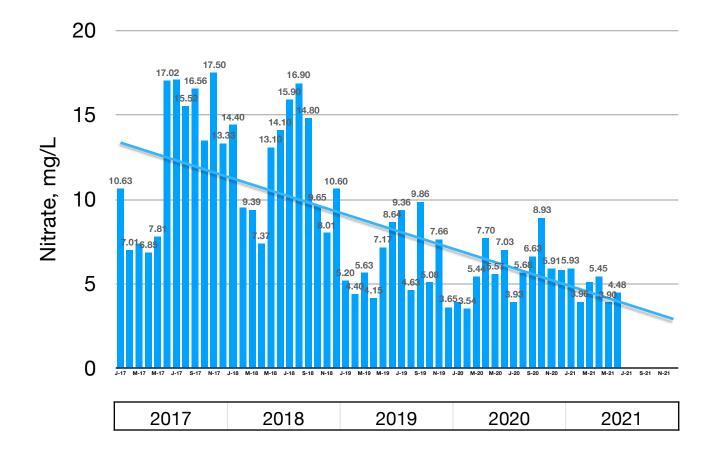
TP load highly variable - seasonal effect & process effect combined

#### Marco Island WWTP - Reuse TKN



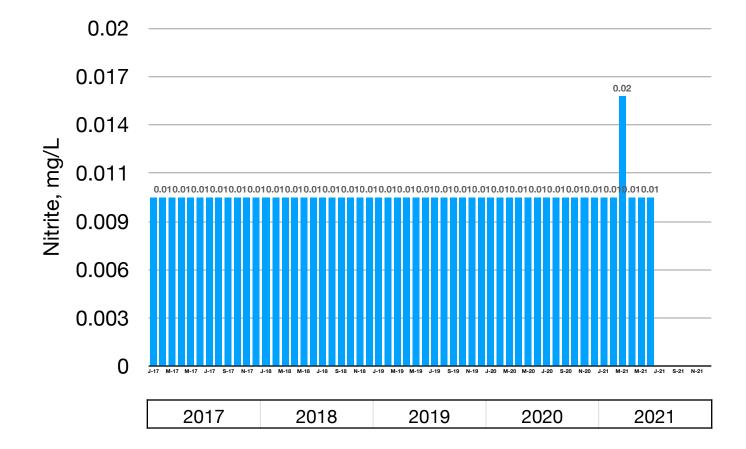
TKN variable and stable

### Marco Island WWTP - Reuse Nitrate (NO3)



Nitrate reduced significantly beginning early 2019

## Marco Island WWTP - Reuse Nitrite (NO2)



Nitrate can not be detected

# **Aquatic Hypoxia**

Oxygen depletion is a phenomenon that occurs in aquatic environments as dissolved oxygen (DO; molecular oxygen dissolved in the water) becomes reduced in concentration to a point where it becomes detrimental to aquatic organisms living in the system. Dissolved oxygen is typically expressed as a percentage of the oxygen that would dissolve in the water at the prevailing temperature and salinity (both of which affect the solubility of oxygen in water; see oxygen saturation and underwater). An aquatic system lacking dissolved oxygen (0% saturation) is termed anaerobic, reducing, or anoxic; a system with low concentration—in the range between 1 and 30% saturation—is called hypoxic or dysoxic. Most fish cannot live below 30% saturation. Hypoxia leads to impaired reproduction of remaining fish via endocrine disruption. A "healthy" aquatic environment should seldom experience less than 80%. The exaerobic zone is found at the boundary of anoxic and hypoxic zones.

Oxygen depletion can result from a number of natural factors, but is most often a concern as a consequence of pollution and eutrophication in which plant nutrients enter a river, lake, or ocean, and phytoplankton blooms are encouraged. While phytoplankton, through photosynthesis, will raise DO saturation during daylight hours, the dense population of a bloom reduces DO saturation during the night by respiration. When phytoplankton cells die, they sink towards the bottom and are decomposed by bacteria, a process that further reduces DO in the water column. If oxygen depletion progresses to hypoxia, fish kills can occur and invertebrates like worms and clams on the bottom may be killed as well.

To combat hypoxia, it is essential to reduce the amount of land-derived nutrients reaching rivers in runoff. This can be done by *improving sewage treatment* and by reducing the amount of fertilizers leaching into the rivers. Alternately, this can be done by restoring natural environments along a river; marshes are particularly effective in reducing the amount of phosphorus and nitrogen (nutrients) in water. Other natural habitat-based solutions include restoration of shellfish populations, such as oysters. Oyster reefs remove nitrogen from the water column and filter out suspended solids, subsequently reducing the likelihood or extent of harmful algal blooms or anoxic conditions. Foundational work toward the idea of improving marine water quality through shellfish cultivation was conducted by Odd Lindahl et al., using mussels in Sweden. More involved than single-species shellfish cultivation, integrated multi-trophic aquaculture mimics natural marine ecosystems, relying on polyculture to improve marine water quality.

Source: Wikipedia